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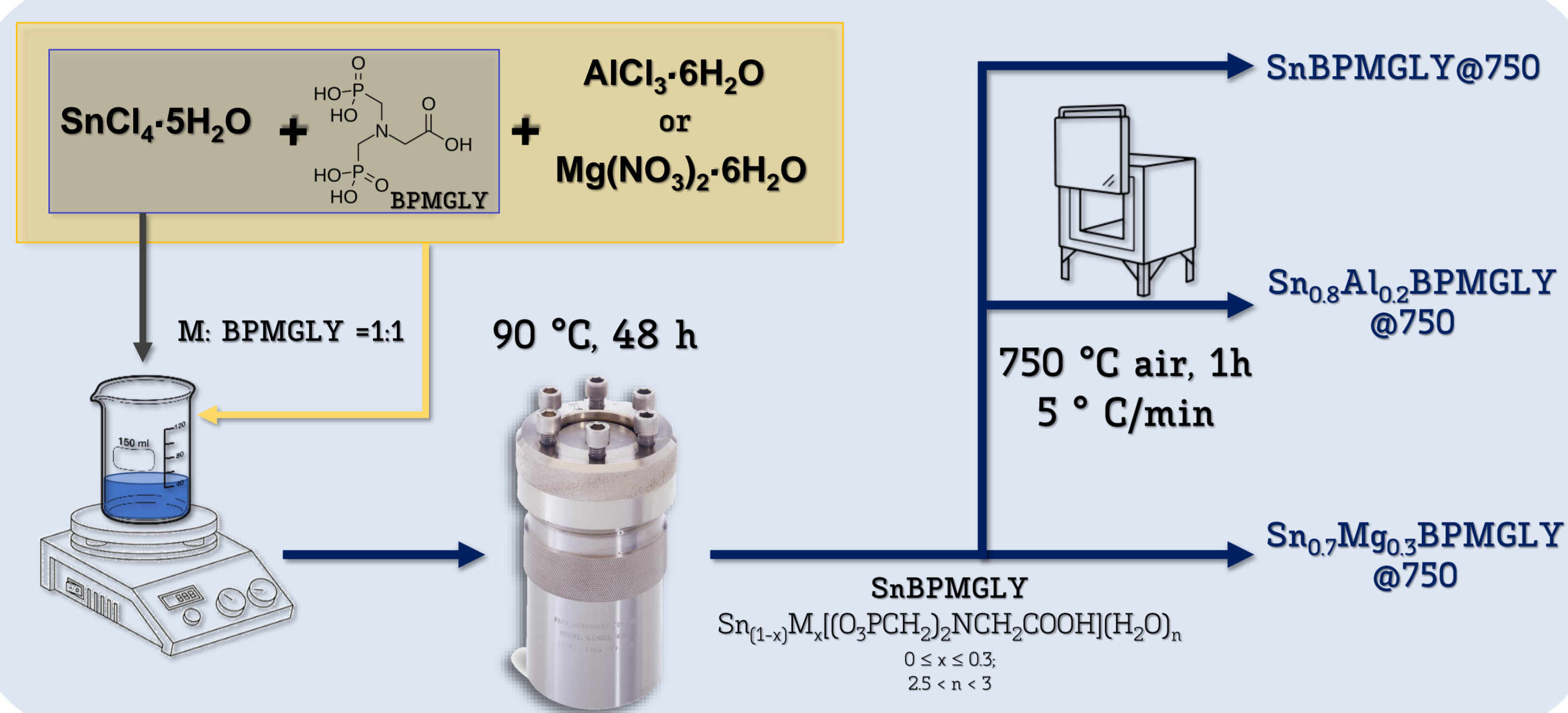
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## Introduction

Metal phosphonates (MPs) combine good chemical and thermal stability alongside tuneable properties, which provide appealing features for electrochemical applications [1-2]. They exhibit not only remarkable proton conductivity properties but also can be potential precursors of electrolytes. useable in intermediate-temperature fuel cells (ITFCs), operating at 100 - 250 °C [3].

In this work, we report the synthesis and characterization of a **new family of amorphous tin(IV) glyphosine-based phosphonates**, Sn<sub>1-x</sub>M<sub>x</sub>[(O<sub>3</sub>PCH<sub>2</sub>)<sub>2</sub>-N-CH<sub>2</sub>CH<sub>2</sub>COOH]<sub>1-x</sub>[(O<sub>3</sub>PCH<sub>2</sub>)<sub>2</sub>-NH-CH<sub>2</sub>CH<sub>2</sub>COOH]<sub>x</sub>·nH<sub>2</sub>O (M = Mg<sup>2+</sup>, Al<sup>3+</sup>; x < 0.3).

## Synthesis



## Pyrophosphate derivatives characterization

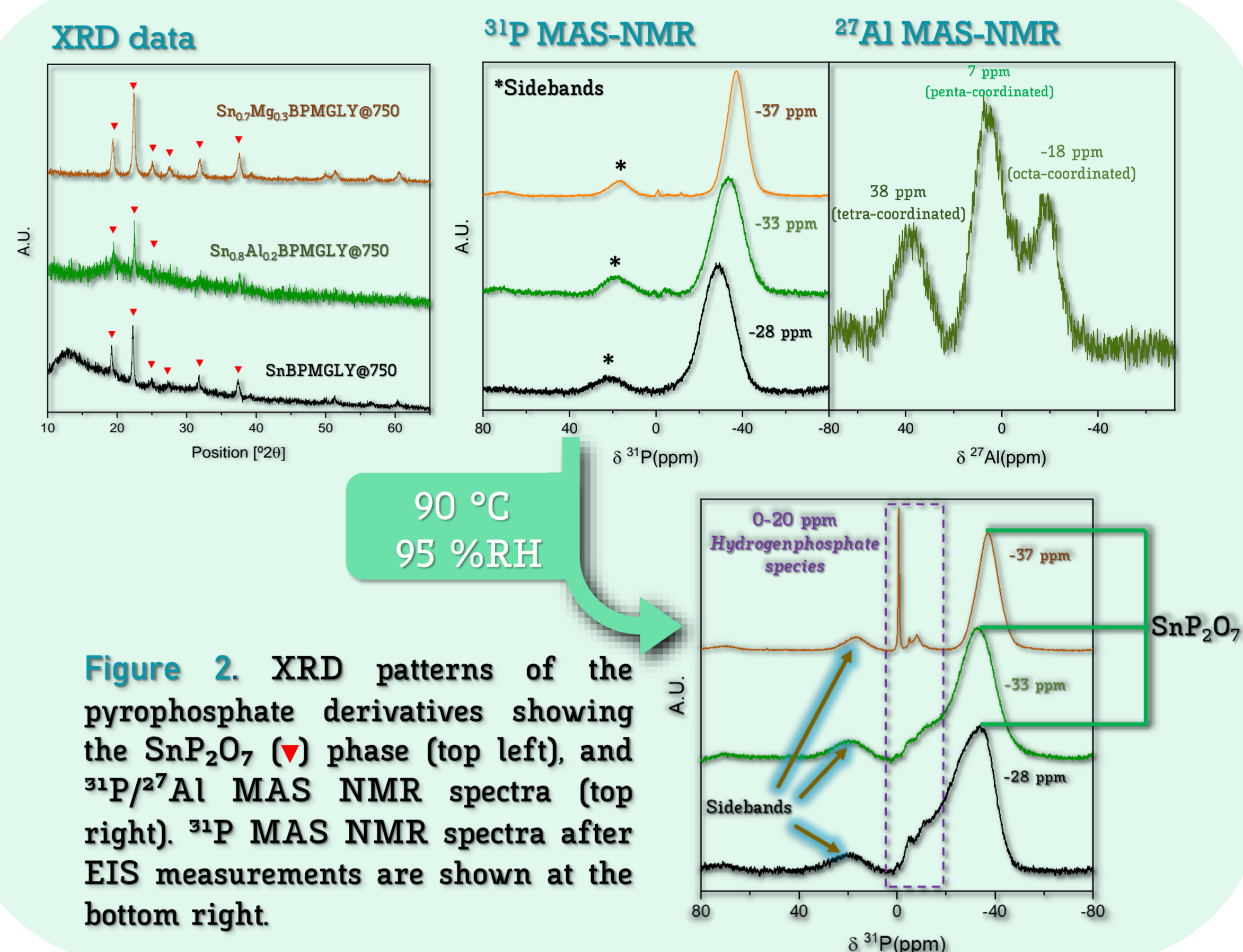


Figure 2. XRD patterns of the pyrophosphate derivatives showing the SnP<sub>2</sub>O<sub>7</sub> (▼) phase (top left), and <sup>31</sup>P/<sup>27</sup>Al MAS NMR spectra (top right). <sup>31</sup>P MAS NMR spectra after EIS measurements are shown at the bottom right.

## Precursors characterization

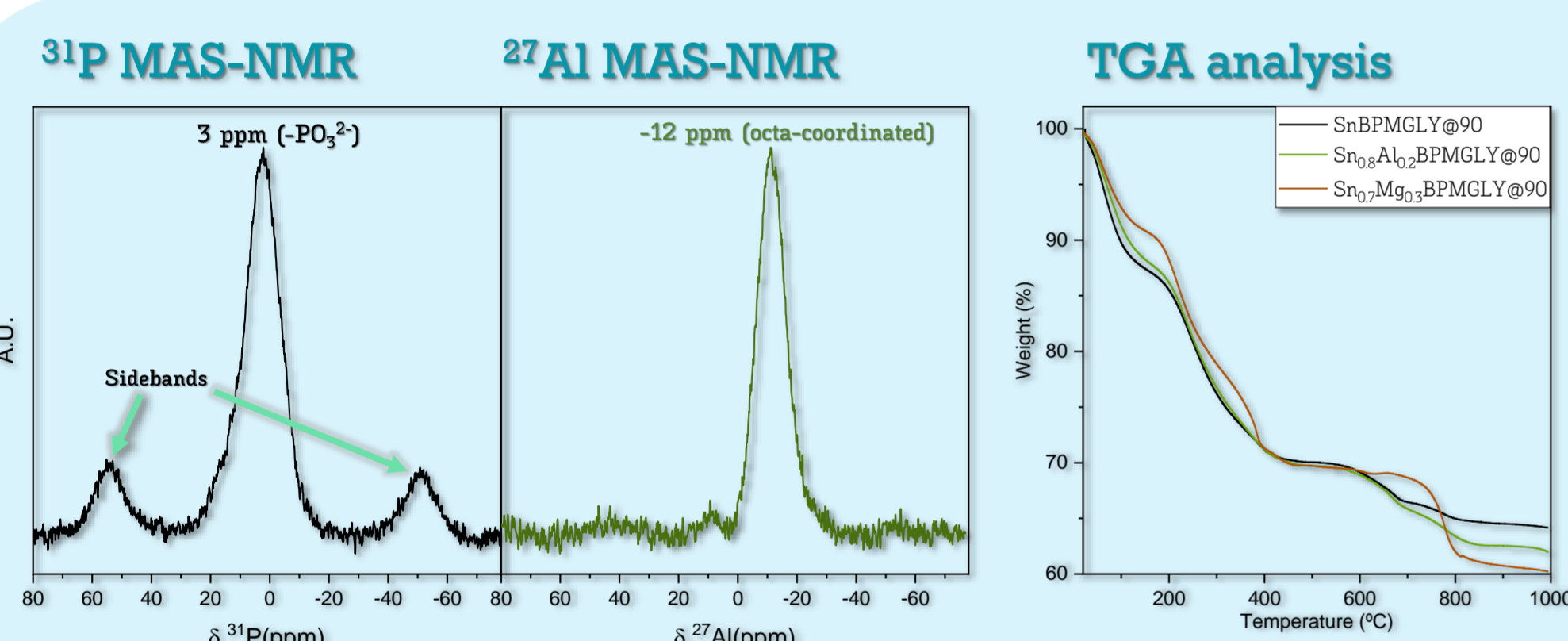


Figure 1. <sup>31</sup>P and <sup>27</sup>Al MAS-NMR spectra for undoped (black) and Al-doped (green) precursors (left) and TGA analysis of the three precursors (right).

## TEM-EDS analysis

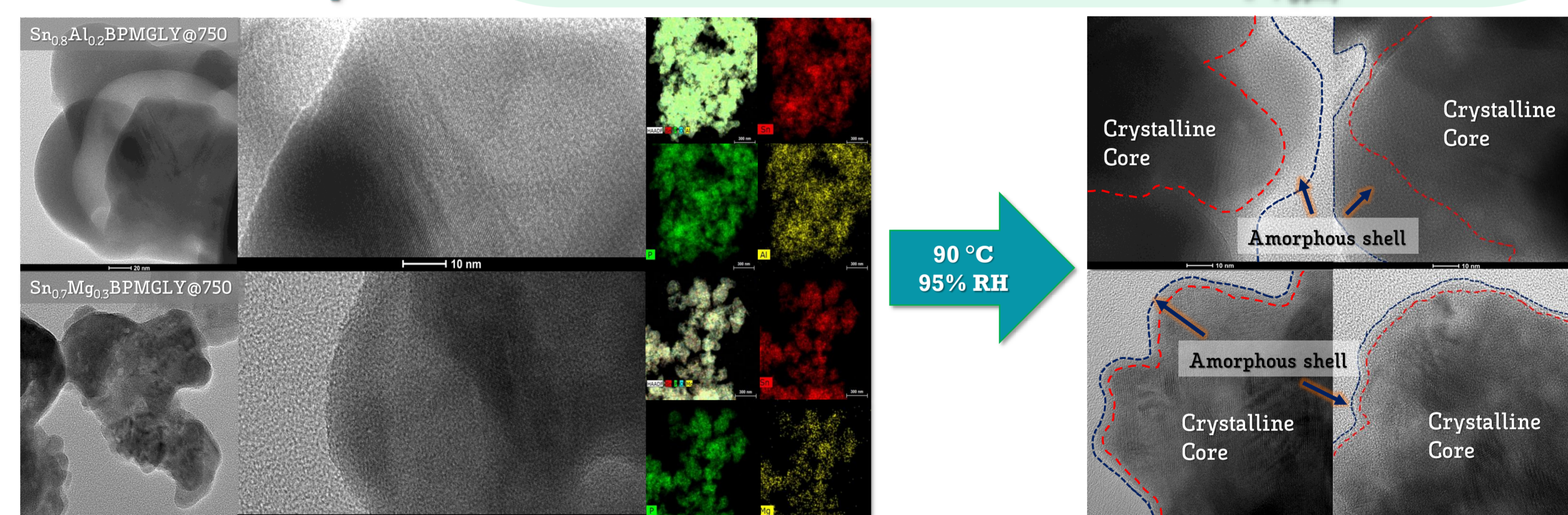


Figure 3. TEM-EDS analysis of both doped pyrophosphate derivatives as well as the study of the core-shell like structure of their particles after EIS measurements

## Proton conductivity properties

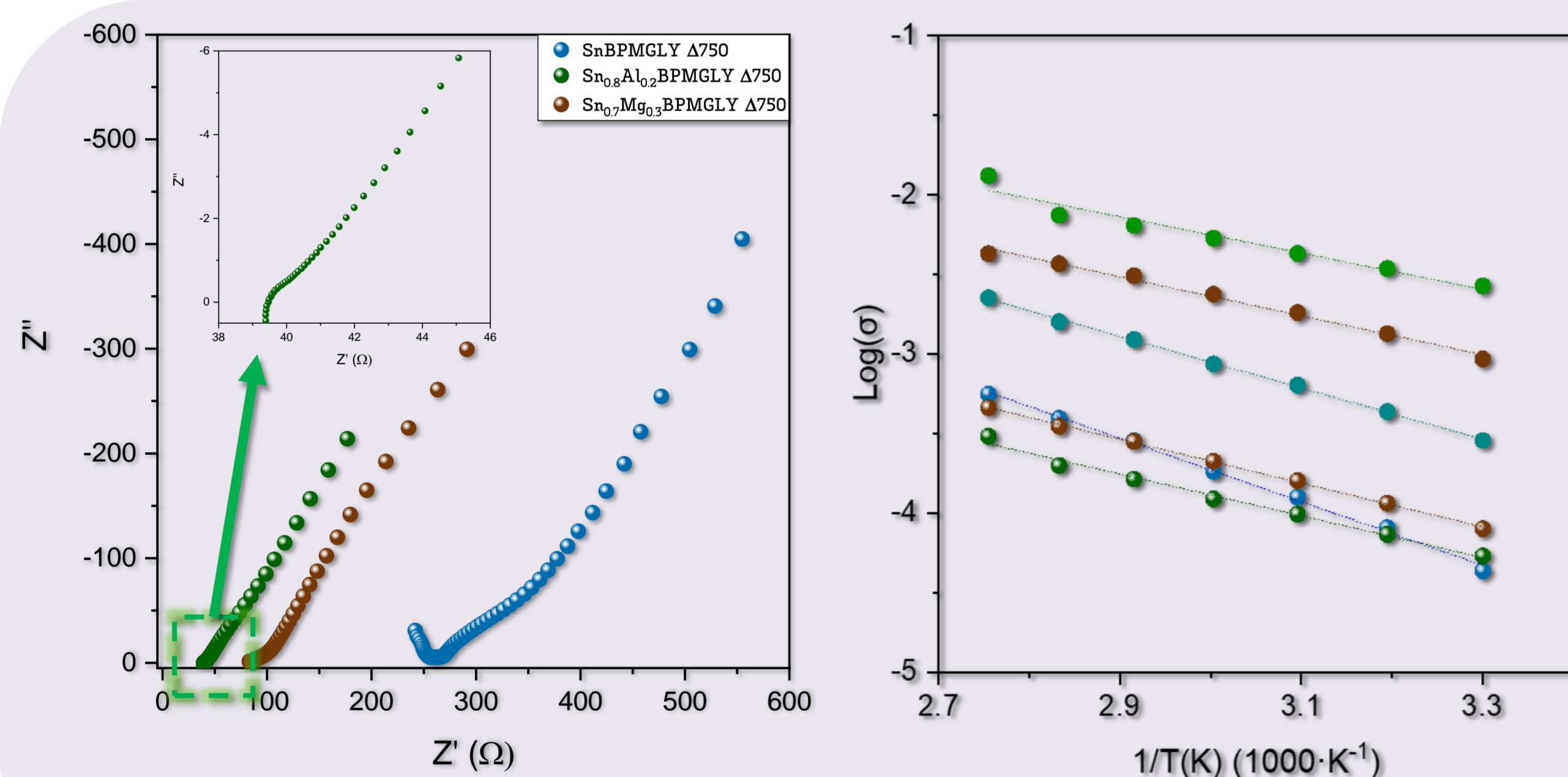


Figure 4. Nyquist plots of pyrophosphates derivatives (left) and Arrhenius plots of all samples.

Sample	σ (S·cm <sup>-1</sup> )		Activation Energies (eV)
	90 °C, 95% RH	90 °C, 75% RH	95% RH
● SnBPMGLY	5.6 · 10 <sup>-4</sup>	3.9 · 10 <sup>-5</sup>	0.326
● Sn <sub>0.8</sub> Al <sub>0.2</sub> BPMGLY	3.0 · 10 <sup>-4</sup>	4.0 · 10 <sup>-4</sup>	0.259
● Sn <sub>0.7</sub> Mg <sub>0.3</sub> BPMGLY	4.6 · 10 <sup>-4</sup>	-	0.243
● SnBPMGLY@750	2.2 · 10 <sup>-3</sup>	4.8 · 10 <sup>-4</sup>	0.316
● Sn <sub>0.8</sub> Al <sub>0.2</sub> BPMGLY@750	1.3 · 10 <sup>-2</sup>	3.3 · 10 <sup>-3</sup>	0.226
● Sn <sub>0.7</sub> Mg <sub>0.3</sub> BPMGLY@750	4.3 · 10 <sup>-3</sup>	1.0 · 10 <sup>-3</sup>	0.243

Table 1. Proton conductivity values of SnBPMGLY precursors and their pyrophosphate derivatives.



## Conclusions

- A new family of amorphous tin(IV) phosphonates has been prepared, characterized and assayed as precursors of tin(IV) pyrophosphate-based proton conductors. The incorporation of doping ions (Al<sup>3+</sup>/Mg<sup>2+</sup>) has improved the proton conductivity up to **1.3 · 10<sup>-2</sup> S·cm<sup>-1</sup>** at 90 °C and 95% RH for **Sn<sub>0.8</sub>Al<sub>0.2</sub>BPMGLY@750**.
- MAS-NMR study showed the presence of hydrogen phosphate groups in all samples after post-impedance measurements, which are apparently responsible for the enhancement of the proton conductivity.
- A **core-shell** structural organization of the materials' nanoparticles was revealed by **TEM**

## Acknowledgments

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## References

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