

Theoretical insights into the Electronic and Structural Properties of New, Low-band Gap Inherently Chiral Ethylenedioxythiophene-based Oligothiophene

Sergio Gámez-Valenzuela^b, Tiziana Benincori^a, Miriam Goll^c, Kirsten Bruchlos^c, Claudia Malacrida^c, Serena Arnaboldi^d, Patrizia Romana Mussini^d, Monica Panigati^d, Juan T. López Navarrete^b, M. Carmen Ruiz Delgado^b, Giulio Appoloni^a, and Sabine Ludwigs^c.

^a Dipartimento di Scienza e Alta Tecnologia, Università degli Studi dell'Insubria, Via Valleggio 11, 22100 Como.

^b Department of Physical Chemistry, University of Málaga, 29071 Málaga (Spain).

^c IPOC-Functional Polymers, Institute of Polymer Chemistry, University of Stuttgart, Pfaffenwaldring 55, 70569 Stuttgart.

^d Dipartimento di Chimica, Università degli Studi di Milano, Via Golgi 19, 20133 Milano.

^e Istituto per lo Studio delle Macromolecole, Consiglio Nazionale delle Ricerche (ISMAC-CNR), Via E. Bassini 15, 20133 Milano, Italy.

In the last years, conjugated oligothiophene macrocycles have attracted increasing scientific interest due to some peculiar properties related to their cyclic structure [1-3]. T. Benincori et al. synthesized the 2,2'-bis(2,2'-bithiophene-5-yl)-3,3'-bithianaphthene nicknamed BT₂T₄ (Figure 1) that represents the first member of a new class of chiral oligothiophenes in which chirality results from a tailored torsion produced in the polyconjugated backbone and not from the presence of stereogenic centres, external to it. Interestingly, the FeCl₃ oxidation of the enantiopure BT₂T₄ produce a mixture of chiral macrocycles, like dimers and trimers. [4]

Recently, also thanks to DFT and TD-DFT calculations, we have studied the new monomer BT₂E₄ in order to investigate the role of the insertion of 3,4-ethylenedioxythiophene (EDOT) units on the electronic and molecular properties of neutral and charged monomer and oligomer species. Furthermore, the electroactive films were evaluated by cyclic voltammetry (CV), UV-vis spectroelectrochemistry and CV coupled with *in-situ* conductance measurements. [5]

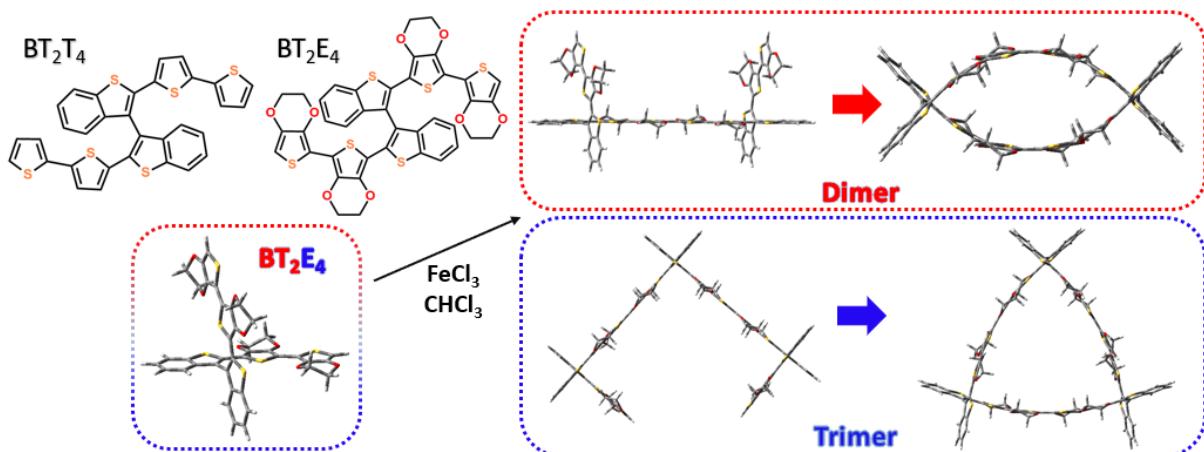


Figure 1 Compounds under study.

[1] F. Zhang, G. Goetz, H. D. F. Winkler, C. A. Schalley, P. Baeuerle, *Angew. Chem., Int. Ed.*, 2009, 48, 6632–6635.

[2] F. Zhang, G. Goetz, E. Mena Osteritz, M. Weil, B. Sarkar, W. Kaim, P. Baeuerle, *Chem. Sci.*, 2011, 2, 781–784.

[3] H. Ito, Y. Mitamura, Y. Segawa, K. Itami, *Angew. Chem., Int. Ed.*, 2015, 54, 159–163.

[4] F. Sannicolo, S. Rizzo, T. Benincori, W. Kutner, K. Noworyta, J.W. Sobczak, V. Bonometti, L. Falciola, P.R.Mussini, M. Pierini. *Electrochim. Acta*, 55 (2010), p. 8352.

[5] T. Benincori, S. Gámez-Valenzuela, M. Goll, K. Bruchlos, C. Malacrida, S. Arnaboldi, P. Mussini, M. Panigati, J. T. López Navarrete, M.C. Ruiz Delgado*, G. Appoloni, and Sabine Ludwigs. *Electrochim. Acta* 284 (2018) 513-525.